



# IN-VITRO FREE RADICAL SCAVENGING AND ANTICANCER POTENTIAL OF METHANOL LEAF EXTRACT OF PLANT *BACOPA MONNIERI* AGAINST HCT15- CELL LINE

Ramachandiran Mangaleshwari  
Department of Zoology,  
Thiruvalluvar University,  
Serkkadu, Vellore-632 115, Tamilnadu, India.

Muthiah Chandran  
Department of Zoology,  
Thiruvalluvar University,  
Serkkadu, Vellore-632 115, Tamilnadu, India.

**Abstract** - It is a universal fact that all living organisms are depending on plants for breathe, food, medicinal and everyday necessities. So, since the period of time immemorial, the peoples are using plant as a medicine to cure various diseases without knowing the scientific back-ground beyond this. The plant *Bacopa monnieri* (L.) Wettst, comes under the family *Scrophulariaceae* are commonly known by different names such as Water hyssop, Baby's tear, Brahmi, Jalbrahmi, Nira-brahmi and Saraswati. In Ayurvedic system of ancient civilization, this plant is used to treat anxiety, improving cognitive functions, memory enhancement, hepato-protection and neuro-protection. This practice is continuing even today in the village side. Hence, the present study has been designed to evaluate the phytochemical constituents, extractive value to determine appropriate solvent to get maximum phytochemicals and antioxidant potential of the plant *Bacopa monnieri*. The obtained results have seen the presence of various important secondary metabolites such as amino acids, steroids, glycosides, flavonoids, alkaloids and tannins in methanolic extract. The Cytotoxicity analysis by the MTT (3-[4,5-dimethylthiazole-2-yl]-2,5-diphenyltetrazolium bromide) assay method showed a significant decline in cell viability in the colon cancer cell line-HCT15 indicates the presence of anticancer activity for this plant extract. The anticancer potential of *Bacopa monnieri* increased with increasing concentrations of plant extract.

**Key words:** *Bacopa monnieri* (L.), free radical scavenging, anticancer, FTIR, GCMS, DPPH, FRAP, nitrous oxide, Phytochemicals

## I. INTRODUCTION

India has a rich culture with good knowledge in usage of medicinal herbs and species. India is one of the 17 megabiodiverse countries in the world. Nearly 50,000-70,000 medicinal plants were reported till date in India which leads to pave a way for very big exposure to traditional medicines such as Ayurvedic, Unani, Siddha from the ancient period. Even today majority of the people in rural India are using plant based traditional medicines to cure the various diseases. But, in this context both the medicinal practitioners and patients are not aware about the chemical constituents of the plant and their disease curing potential. Till date a very few studies were conducted to identifying phytochemical and pharmacological potential to cure the diseases [1-2]. Recently the most of the researchers are involving in phytochemical study on plants to know their phytochemical constituents, properties and various diseases [3]. Usually, in human being vast majority of diseases are caused by the free radical generation during metabolic processes. The natural defense of the human body against free radicals is not always sufficient mainly due to the significant exposure to pollutant substances. But, the medicinal plants having the rich antioxidants are very helpful to balance the accumulation of free radicals in human body. The plant *Bacopa monnieri* (L) listed the family Scrophulariaceae, has been used for more 3000 years as Indian Ayurvedic medicines for improving memory, increasing brain function, or promoting longevity [4-6]. *Bacopa monnieri* is a small creeping plant having numerous branches, fleshy, oblong leaves and small plant. Fruits and flowers are appearing in summer period the whole plant is medicinally important [7]. It is also possesses anti-inflammatory, analgesic,



antipyretic, epilepsy, insanity, anticancer and antioxidant activities [8-11]. In addition, the plant has been recommended as an agent for phytoremediation [12-15]. Hence, the in present investigation has been programmed to evaluate the qualitative of phytochemical analysis, extractive value and antioxidant potential of *Bacopa monnieri*.

## II. MATERIALS AND METHOD

### A. Collection of plant material and authentication-

The whole plant of *Bacopa monnieri* (L) was collected from Nowlock garden, Avaraikkarai village, Ranipet area, Vellore District, Tamil Nadu, India. The plant sample was identified and authenticated by Institute of herbal science, Plant anatomy research Centre, Chennai, Tamil Nadu. The identification specimen number is (PARC\17\3535). The leaves were alone handpicked from plant and washed with distilled water to remove dust and other small insects stick on it. The cleaned leaves were dried for 30 days at room temperature (32<sup>0</sup>C) in shade place to avoid the loss of vitamins and denature of phytochemicals. After the completion of drying process, the shade dried leaves were powdered coarsely using electric blender. Then the powder was stored into air tight container for future uses.

### B. Determination of Extractive value-

The dry *Bacopa monnieri* plant powder was extracted in different solvents such as water, methanol, ethanol, ethyl acetate, chloroform, petroleum ether and hexane by using the maceration process. For this study, accurately 1gm of plant powder was taken in a small beaker (100ml) and 50ml of any one of above mentioned desired solvent was mixed with it. The entire setup was kept at room temperature for 24 hours. The mixture was shaken frequently. Thereafter, the extract was filtered using No.1 Whatmann filtered paper [16]. The filtrate was shifted to pre-weighed petri plates. Date and weight of the plate it were marked. The extract was kept upto complete evaporation of solvent. Finally, the dried extract containing petri plates was weighed. The extractive value (%) was calculated used by following formula.

**Extractive value (%) =** Weighed dried extract/ Weigh of plant material X 100

### C. Preparation of the plant extract

*Bacopa monnieri* plant powder 20gm was packed in a thimble and 200 ml of methanol solvent was filled in the bottom of soxhlet. The heating mantle was set at 60<sup>0</sup>C [17]. After the completion of extraction, the methanolic leaf extract collected on the bottom were concentrated by evaporating it to dryness under reduced pressure by

rotary vacuum evaporator to obtain the dried sample. The sample was stored at 180<sup>0</sup>C for further analysis.

### D. Preliminary Qualitative phytochemical determination

The methanolic plant extract obtain from soxhlation process were subjected to preliminary qualitative determination for the presence of phytochemicals such as cardio glycosides, alkaloids, flavonoids, saponins, phenols, steroids, Proteins, Anthraquinones, coumarin, triterpenes, quinines and tannins by standard methods [18].

#### i. Test for tannins:

The methanol leaf extract of the plant *Bacopa monnieri* was taken in three separate test tubes. In each test tube few drops of 0.1 % ferric chloride were added. The brownish green or blue black colour was formed in test tube indicates the presence of tannin.

#### ii. Test for saponins:

1 ml of leaf extract and 2 ml of water was taken in a test tube. This mixture was shaken 15 minutes. A layer of foam is appeared. The appearance of foam layer indicates the presence of saponins.

#### iii. Test for flavonoids:

1 ml of methanolic plant leaf extract was taken in test tube. Addition to this 1ml of NaoH was added. Now, the yellow colour formation in test tube formed into white colour after adding few drops of con. H<sub>2</sub>SO<sub>4</sub> confirmed the presence of flavonoids.

#### iv. Test for alkaloids:

##### Drangandroff reagent

1 ml of *Bacopa monnieri* plant extract was taken along with, few drops of drangandroff reagent was added. A prominent yellow colour precipitates was formed in a test tube. It indicates presence of alkaloids as positive.

##### Mayer's test

Take 1ml of plant extract in a test tube. Along with few drops of Mayer's reagent were added. The white creamy was observed indicates the presence of alkaloid.

##### Wagner's test

Methanolic leaf extract was taken in a test tube, along with few drops of Wagner's reagent were slowly added. Creamish brown colour precipitate in a test tube confirmed the presence of alkaloids.

#### v. Test for protein:

1 ml of sample was taken. Along with few drops of Millon's reagent was added. Appearing of white



precipitate in the test tube indicates the presence of Protein.

**vi. Test for steroids:**

1 ml of *Bacopa monnieri* plant extract was taken in a test tube, along with two drops of 10% concentrated sulphuric acid was added and observed for brown colour. Formation of characteristic the brown colour precipitate indicates the presence of steroids.

**vii. Test for anthraquinones:**

1 ml of plant extract was taken in a test tube along with aqueous ammonia solution was added and observed for colour changes in a test tube. Appearing of pink colour in aqueous layer confirmed the presence of anthraquinones.

**viii. Test for phenols:**

1ml crude plant extract was taken in a test tube. After that 3ml of 10% lead acetate solution was added. The bulk white colour precipitate formed at the surface indicates the presence of phenolic compounds.

**ix. Cardio glycosides:**

1 ml of plant extract was taken in a test tube, along with 1ml of glacial acetic acid, 1 ml of ferric chloride few drops of concentrated sulphuric acid was added. The appearance of reddish brown at the junction of the 2 liquid layers indicates presence of cardio glycosides.

**x. Coumarin:**

2 ml of 10% sodium hydroxide was added to 2ml of plant extract. Formation of yellow colour in the test tube indicates presence of coumarins.

**xi. Triterpenes:**

2ml of methanolic leaf extract was taken along with few drops concentrated sulphuric acid was added. The formation of reddish brown ring in a test tube indicates the presence of Triterpenes.

**xii. Quinines:**

1ml of plant extract was taken in a test tube, along with 1ml of 1% sodium hydroxide were added and mixed well. The appearance of dark blue green indicates confirmed the presence of quinines.

**D. FTIR analysis**

Functional groups of phytochemicals present in the plant were identified by using the powerful tool FTIR. The FTIR model Jasco was used in this study. For this study 5mg of dried fine plant powder was taken and mixed with 50mg of kbr to make a pellet. This pellet was loaded in the FTIR and scan at room temperature in the spectral ranges from 400-4000  $\text{cm}^{-1}$ . The results obtain from the

FTIR were interpreted by analysed the chemical bonds for particular peak value.

**E. TLC**

5gram dried leaf powder of *Bacopa monnieri* was taken in a conical flask. Along with 10ml of methanol was added. This setup was kept in magnetic sterrier for 1 day to obtain complete dissolving of phytochemicals of leaf in the methanol solvent. The extract obtained from this maceration process was filtered through No.1. Whattmann filter paper. The appropriate solvent system chosen for the present study was composed of methanol: chloroform: hexane (14:4:2). The silica gel precoated readymade aluminum plate obtained from Harish Scientific Company in Vellore used as a stationary phase. Thereafter 5 $\mu$ l of sample was spotted on the marked place on TLC plate and allowed to run in a TLC chamber saturated with the mobile phase of solvents. Then the spot of antioxidant activity was qualitatively screened by dipping the TLC plates down for 10seconds in 0.5mm in DPPH solution. The chromatogram of TLC turned into purple colour due to the presence of antioxidant activity.

**F. Antioxidant assays**

**i. Determination of DPPH scavenging assay:**

The DPPH (2,2-diphenyl-1-picrylhydrazyl) free radical scavenging antioxidant activity of plant extract was determined by the method followed Blois [19]. An aliquot amount of 0.5 ml methanolic leaf extract of *Bacopa monnieri* was taken in a test tube, along with 2.5 ml of 0.5 mm methanolic solution of DPPH was added. The sample mixture was shaken vigorously and incubated for 30 min in the dark place at room temperature. Thereafter the colour of the solution was measured by using UV spectrophotometer at 517nm (triplicate values). The ascorbic acid was used for a positive control. The scavenging activity of the plant extract was calculated by using the following formula and expressed in a unit % of inhibition.

**% of inhibition** =  $\frac{\text{Absorbance of control} - \text{Absorbance of sample}}{\text{Absorbance of control}} \times 100$ .

**ii. FRAP Assay**

The ferric reducing power of *Bacopa monnieri* methanolic leaf extract were determined by using FRAP method [20-21]. This reaction is based on the reduction of colorless ferric complex ( $\text{Fe}^{3+}$  tripyridyltriazine) to blue-colored ferrous complex ( $\text{Fe}^{2+}$  tripyridyltriazine) by the action of electron donating antioxidants at low pH. The reduction was monitored by measuring the change of absorbance at 593 nm. The working FRAP reagent was prepared by mixing 10 volumes of 300 mM acetate buffer, pH 3.6, with 1 volume of 10mM TPTZ (2,4,6-tri(2-pyridyl)-striazine) in 40m Mm HCl and with 1



volume of 20 mM ferric chloride. All the required solutions were freshly prepared just before use. 100  $\mu$ L of samples (mg/mL) were added to 3mL of prepared FRAP reagent. The reaction mixture was incubated in a water bath for 30min at 37°C. The standard curve of ferric chloride (125  $\mu$ mol, 250  $\mu$ mol, 500  $\mu$ mol, 750  $\mu$ mol and 1000  $\mu$ mol) was prepared using by same procedure. Then, the absorbance of the samples was measured at 593 nm. The difference between absorbance of sample and the absorbance of blank were calculated to determine the FRAP value. FRAP value was expressed in terms of Mmol Fe<sup>2+</sup>/g of sample using ferric chloride standard curve  $Y = 1.7057x - 0.2211$ ,  $R^2 = 0.9904$ .

### iii. Nitric oxide radical scavenging activity

The nitric oxide radical scavenging was measured by Griess reaction [22]. For this experiment, the reaction mixture (3ml) containing sodium nitroprusside 10mM in phosphate buffered saline was mixed with various concentrations of methanolic plant extract and incubated at 25°C for 2.5 hours. After the completion of incubation, 0.5ml aliquot amount of Griess reagent [1% sulfanilamide, 2% H<sub>3</sub>PO<sub>4</sub> and 0.1% N- (1-naphthyl) ethylene diamine dihydrochloride] was added. Finally, the chromophore formed in the reaction was measured immediately at 546nm. The inhibition of nitrate formation by the plant extract and the standard antioxidant ascorbic acid were calculated relative to the nitric oxide radical control. The experiment was triplicated and the percentage of scavenging activity was estimated with Curcumin standard reference.

### G. Cytotoxicity of methanol leaf extract of *Bacopa monnieri* on Colon cancer cell line-HCT15

Cytotoxicity analysis was done by the MTT (3-[4,5-dimethylthiazole-2-yl]-2,5-diphenyltetrazoliumbromide) assay method. Working principles of this method is based on the conversion of MTT into formazan crystals by living cells contain NAD(P)H- dependent oxidoreductase enzyme. This assay technology has been widely adopted and remains popular in academic labs to find homogeneous cell viability. The MTT substrate is prepared in a physiologically balanced solution, which has added to culture at a final concentration of 0.2-0.5mg/ml, and incubated for 1 to 4 hours. The formazan crystals formed in the cell culture were then dissolved using solubility solution and absorbed at 570 nm using a plate reading spectrophotometer. The darker solution means the greater the viable and metabolically active cells. The Colon cancer cell line (HCT15) was placed in 96 well plates containing DMEM media with 1X Antibiotic Antimycotic solution and 10% fetal bovine serum (Himedia, India). Along with different concentration of methanol leaf extract of *Bacopa monnieri* were added. Before that, the concentration of

cells was determined to  $1 \times 10^7$  cells/well. Then the cell line was incubated in CO<sub>2</sub> incubator at 37° with 5% CO<sub>2</sub>. After incubation, the cell line was washed with 200 $\mu$ l of 1X PBS, and MTT reagents were added to each well. The MTT reagent 5 ml containing 5 vials purchase from Sigma Company was used for this experiment. In the present study, this MTT assay method was used to measure the cytotoxicity of plant extract on necrosis of cancer cells in colon.

## III. RESULTS & DISCUSSION

The extractive value for *Bacopa monnieri* leaf was estimated using different solvents and the obtained results were recorded in Table-2 and depicted in Fig.1 (a and b) which showed that the highest yield of extractive value obtained for both methanol (4.9gm) and water (3.6gm) compared all six solvents. The extractive values of other solvents were observed following sequences ethanol (2.2gm) > ethyl acetate (2gm) > chloroform (0.9gm) > hexane (1.7gm) > petroleum ether (0.6gm). Hence the high extractive methanol solvent was selected to extract the phytochemicals from *Bacopa monnieri*. This result showed a concordance with the findings suggested by [23] that the solvents methanol and acetone appeared to be more effective in extracting bioactive compounds from *Mentha viridis*. 50% acetone is an ideal solvent to extract the phenol from *Salacia chinensis* [24].

Plant based antioxidants play a vital role even in trace amount, is capable of preventing of delaying the oxidation of easily oxidizable materials and emerged as a potential therapeutic to prevent free radical generated damage in the human body [25], flavonoids and phenolic [26]. It acts as reductants and inactivators of oxidants [27]. Most of the plants traditionally used to treat various because of having rich antioxidant which are in the form of phytochemicals such as tannins, flavonoids, alkaloids, steroids, etc. Flavonoids present in dietary sources such as tea, red wine, apple tomato, cherry, anion, thyme, parsley, soya-beans and other legumes in the form of flavonols, flavones, isoflavones and flavonones [28]. The preliminary qualitative phytochemical analysis of leaf of methanol extract of plant *Bacopa monnieri* has been exhibited the presence of various bioactive compounds such as tannins, saponins alkaloids, steroids, flavonoids proteins, cardioglycolysis and triterpens (Table-1) as similar as with the qualitative analysis of the plant *Ephedra intermedia* showed the phytochemicals cardioglycolides, alkaloids, reducing sugars, flavonoids, phenols (Ethanol, methanol) [29], phenols and saponins in the methanol extracts of leaves and flowers; alkaloids, flavonoids, terpenoids, carbohydrates, protein and amino acids are present in methanolic extract of leaves, roots and flowers of





*Moringa concansis* [30], steroids, alkaloids, cyanide-I mild concentrations in *Senna mimosoides*. Saponin, phenols, flavonoids, terenoids, soluble carbohydrate and tannins [31]; carbohydrate, amino acids, phenols and alkaloids in the flowers of *Bauhinia acuminata* [32]; alkaloids, terpinoids, steroids, flavonoids, polyphenols, glycosides, tannin and saponins in *Acanthopora spicifera* and *Sargasum wightii* [33]; alkaloids, terpenoids, steroid, tannins, flavonoids, phenols, coumarins, quinons and glycosides in the brown seaweed *Dictota dichotoma* [34]; tannins, saponins, flavonoids, cardioglycides and alkaloids in the crude extract of *Ephedra altissima* aqueous and extract of leaf of *Ephedra altissima* [35]; cardioglycides, alkaloids in the stem and fruit, flavonoids present in the leaf of *Cissus quadrangularis* [36]. After confirmation, the free radical scavenging ability of this plant leaves were determined by observing DPPH, FRAP, Nitric oxide scavenging activities. Even though a number of methods have been proposed to determine the antioxidant activity, in the present investigation a widely used method [37-38] is used to evaluate the free radical scavenging activity and antioxidant activities of plant extracts.

The leaf extract of *Bacopa monnieri* has the ability to scavenge 72.38% of DPPH at 250mg concentration as similar as exhibited (71.18%) in methanol extracts of the stem of *C. africana* (89.69%) and leaves (64.95%) at 0-1 µg/ml and 82.58 % of activity at 500 µg/ml [39]. The overall observation of DPPH scavenging activities in the leaf powder of *Bacopa monnieri* has an increasing trend when concentration is increase. Therefore, the free radical scavenging activities of *Bacopa monnieri* leaf directly proportionate to their concentration. Nitric oxide (NO) is an important chemical mediator generated by endothelial cells, macrophages, neurons, etc. which is involved in the regulation of various physiological processes [40]. NO is generated in biological tissues by specific nitric oxide synthesis (NOSs), which metabolizes arginine to citrulline with the formation of NO via a five electron oxidative reaction [41]. Nitric oxide or reactive nitrogen species, formed during their reaction with oxygen or with superoxides, such as NO<sub>2</sub>, N<sub>2</sub>O<sub>4</sub>, N<sub>3</sub>O<sub>4</sub>, NO<sub>3</sub> and NO<sub>2</sub> are very reactive. These compounds are responsible for altering the structural and functional behavior of many cellular components. The experimentation on scavenging activity of phytocompounds of *Bacopa monnieri* is very important because of its neuroprotection activities. Fresh leaf extract of *Ixora coccinea*, showed 49.43% and 72.18% NO scavenging, dry flower extract showed 59.43% and 66.93% of NO inhibition and fresh flower has 69.45% and 77.24% of NO inhibition. FRAP assay was used by several authors for the assessment of antioxidant activity of various food product samples [42-43]. Like DPPH and

Nitric oxide, the same trend of FRAP scavenging activities was observed (Ferric reducing ability of plasma), (Table-3 Fig.2(a, b and c). This results showed some similarity with the antioxidant activities of five *Salvia* species such as *Salvia multicaulis*, *Salvia verticellata*, *Salvia lechnecelyx*, *Salvia mirzayanii* and *Salvia macrosiphon* [44].

The FTIR analysis made on methanol, ethanol, hexane, chloroform and petroleum ether leaf extract of *Bacopa monnieri* have 11, 18, 12, 14 and 19 peaks respectively. The extraction of leaf of *Bacopa monnieri* in all solvents such as ethanol, hexane, chloroform and petroleum ether except the methanol showed bands at 430.26(ethanol), 433.50(hexane), 420.56(Chloroform) and 407.62, 72.32 and 549.96(petroleum ether) cm<sup>-1</sup> with C-I stretching indicate the presence of halo compounds. The band at 572.60 (methanol), 769.94(ethanol), 608.19(chloroform,) 653.48 and 889.64(petroleum ether) would be due to C-Br stretch, C-H bend indicate the presence of Alkyl halides and Alkynes. The band absorbed at 931.70(methanol), 902.58(ethanol), 909.05(hexane), 931.70(Chloroform) and 993.17(petroleum ether) due to O-H bend of carboxylic acids. 1122.57(methanol), 1035.22, 1187.27 and 1264.91(ethanol), 1077.28 and 1287.56(hexane), 1064.34 and 1196.98(Chloroform) and 1109.63(petroleum ether) due to C-N stretch indicate the presence of aliphatic amines. 1407.26 and 1501.07(methanol), 1339.32, 1452.55 and 1530.19(ethanol), 1420.20 and 1497.84(hexane), 1309.26, 1393.31 and 1501.07(Chloroform) and 1326.38, 1404.02 and 1478.43(petroleum ether) due to N-O symmetric stretch, C-C stretch (in-ring) indicate the presence of Nitro compounds, aromatics. 1653.12(methanol), 1624.01(ethanol), 1633.71(Chloroform) and 1649.89(petroleum ether) due to N-H bend indicate the presence of amines. 1824.58(methanol), 1701.65 and 1841.88(ethanol), 1782.53(hexane), 1824.58(Chloroform) and 1772.82(petroleum ether) due to C=O bend and indicate the presence of Esters, Anhydrides. 2034.86(methanol) due to C≡C Stretch indicate the presence of Alkynes (monosubst). 2856.57(methanol), 2914.80(ethanol), 2882.45 and 2940.69(hexane), 2924.51(Chloroform) and 2798.34, 2843.63 and 2930.98(petroleum ether) due to CH<sub>3</sub> and CH<sub>2</sub>, C-H stretch indicate the presence of Alkanes, Aliphatic compounds. 3086.26(ethanol), 3015.09(hexane), 3002.15(Chloroform) and 3015.09(petroleum ether) due to =C-H stretch indicate the presence of Alkenes. 3456.57(methanol), 3429.18 and 3464.77(ethanol), 3358.01(hexane), 3419.48(Chloroform) and 3432.42(petroleum ether) due to N-H stretch, O-H stretch indicate the presence of Amines or Alcohols. 3875.62(methanol), 3940.33 and 3998.56(ethanol), 3946.80(hexane),



3742.99(Chloroform) and 3742.99 and 3979.15(petroleum ether) due to -OH stretch indicate the presence of Alcohols and Phenols.

The screening of plant extract for cytotoxicity on colon cancer cell line of In-vitro study showed a significant level of decline in cell viability (Table.4 and Fig 3). These declining trends were increased in increasing concentration. This means the death of colon cancer cells are increasing with increasing concentration of plant extract which indicate the leaf extract of *Bacopa monnieri* possess potent anticancer activities. The present study results showed a strong affinity with *Rosa anina* extract exhibited a selective cytotoxic effect on colon cancer cells compared with normal colon cells. The extract induced cell cycle arrest at the S-phase and apoptosis via reduced MMP in WiDr cells. *Rosa canina* extract significantly repressed telomerase expressions at treatment times of 48 and 72 h in WiDr cells [45]. Aerial, leaf and stem extracts of plant *Alternanthera sessilis* greatly suppressed the growth of colon cancer cells in time and dosage-dependent manner. The cytotoxicity results were rationalized with clonogenic, cell motility and AO/PI assay, where the leaf extract showed the most active activity compared to aerial and stem extracts. [46]. The ethanolic root extract of *Euphorbia tehranica* showed a significant cytotoxic effect against Caco-2-cell line ( $p \leq 0.05$ ). The viability of Caco-2 cells reduced with the dose and time dependent manner [47].

**Table-1. List of phytochemicals present in solvent extract of *Bacopa monnieri***

S.No	Phytochemical tests	Methanol
1.	Tannins	+
2.	Saponins	+
3.	Phenols	-
4.	Alkaloids	+
5.	Coumarin	-
6.	Steroids	+
7.	Anthraquinones	-
8.	Flavonoids	+
9.	Cardio glycosides	+
10.	Proteins	+
11.	Triterpenes	+
12.	Quinines	-

**Table-2. The extractive value of using different solvents and their dry weight *Bacopa monnieri***

Solvents	Dry weight of raw leaf powder	% of extractive value
Water	1	3.6
Methanol	1	4.9
Ethanol	1	2.2
Petroleum ether	1	0.6
Chloroform	1	0.9
Hexane	1	1.7
Ethyl acetate	1	2

**Table-3. Free radical scavenging activities of *Bacopa monnieri* at different concentration**

Free radical scavenging activities			
Different concentration of leaf extract mg	DPPH	FRAP	Nitric oxide
50	23.89±0.81	34.76±0.17	12.31±0.81
100	39.47±0.5	40.05±0.02	27.45±0.98
150	59.98±0.69	42.43±0.06	41.62±0.98
200	66.16±0.50	46.78±0.07	62.09±1.31
250	72.38±0.56	51.50±0.02	73.42±0.82

**Table-4. Colon cancer cell line different concentration of *Bacopa monnieri***

Tested concentration(µg/ml)	% of cell viability
25	97.74 ± 2.25
50	90.21 ± 2.33
100	79.57 ± 1.59
250	67.63 ± 3.18
500	53.33 ± 3.42
Control	100

**Table-5. FTIR analysis of the plant *Bacopa monnieri* in different solvents**

Solvents					Bonding	Functional Group
Methanol	Ethanol	Hexane	Chloroform	Petroleum ether		
-	430.26	433.50	420.56	407.62 472.32 549.96	C-I stretching	Halo Compounds
572.60	769.94		608.19	653.48 889.64	C-Br stretch, C-H bend	Alkyl Halides, Alkynes,
931.70	902.58	909.05	931.70	993.17	O-H bend	Carboxylic Acids
1122.57	1035.22 1187.27 1264.91	1077.28 1287.56	1064.34 1196.98	1109.63	C-N stretch	Aliphatic Amines
1407.26 1501.07	1339.32 1452.55 1530.19	1420.20 1497.84	1309.26 1393.31 1501.07	1326.38 1404.02 1478.43	N-O symmetric stretch, C-C stretch (in-ring)	Nitro Compounds, Aromatics
1653.12	1624.01		1633.71	1649.89	N-H bend	Amines
1824.58	1701.65 1841.88	1782.53	1824.58	1772.82	C=O	Esters, Anhydrides
2034.86					C≡C Stretch	Alkynes (Monosubst)
2856.57	2914.80	2882.45 2940.69	2924.51	2798.34 2843.63 2930.98	CH <sub>3</sub> and CH <sub>2</sub> , C-H stretch	Alkanes, Aliphatic Compounds
	3086.26	3015.09	3002.15	3015.09	=C-H stretch	Alkenes
3456.57	3429.18 3464.77	3358.01	3419.48	3432.42	N-H stretch, O-H stretch	Amines or Alcohols
3875.62	3940.33 3998.56	3946.80	3742.99	3742.99 3979.15	-OH stretch	Alcohols and Phenols

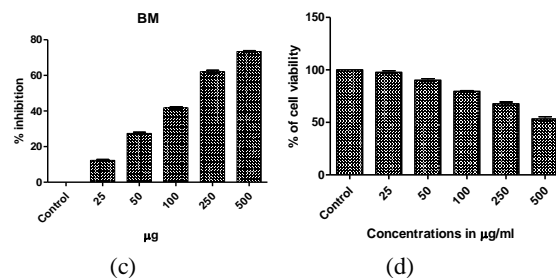


Fig. 2. (a) DPPH in different concentration (b) FRAP in different concentration (c) NO in different concentration (d) Colon cancer in different concentration

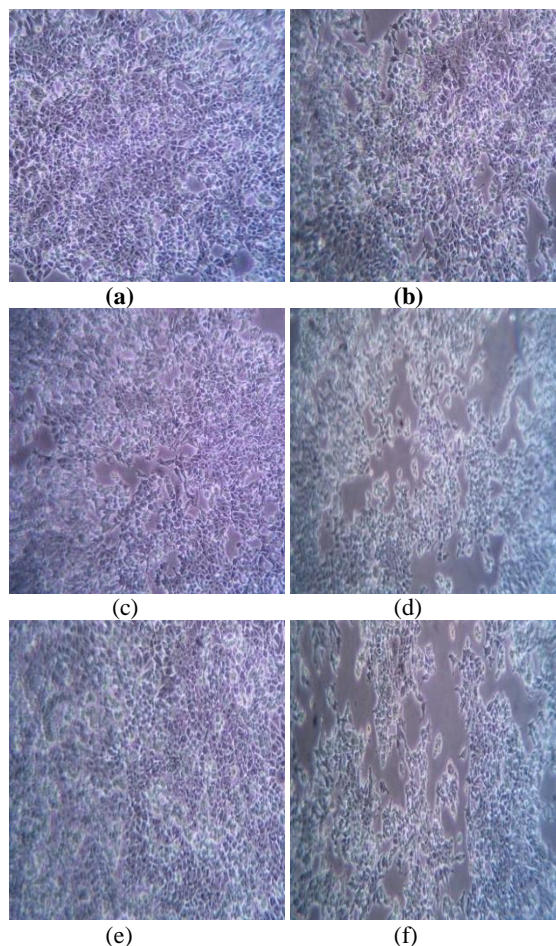


Fig.3. (a) Colon cancer cell line (a) 25 µg/ml (b) 50 µg/ml (c) 100µg/ml (d) 250µg/ml (e) 500 µg/ml (f) Control

#### IV. CONCLUSION

The present study was concluded that the methanolic plant extract of *Bacopa monnieri* having a high potential free radical scavenging and anticancer activity due to the presence of phytochemical contents such as alkaloids,

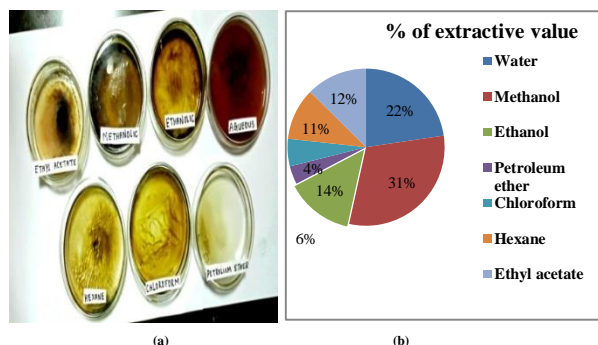
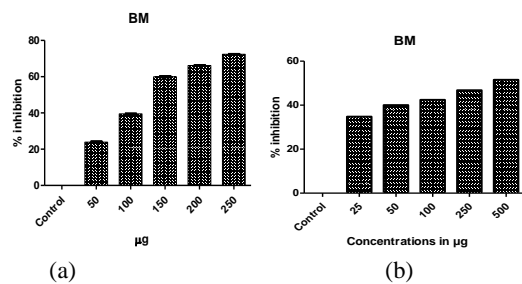


Fig. 1. (a) Dried plant extract solvent (b) % of extractive value







terpenoids, steroid, tannins, flavonoids, phenols, coumarins, quinones and glycosides.

#### V. REFERENCE

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