UTILIZATION OF SOLID LIPID NANOPARTICLES LOADED ANTICANCER AGENTS AS DRUG DELIVERY SYSTEMS FOR CONTROLLED RELEASE

Ibtihag Yahya  
Research Assistant  
Department of Biomedical Engineering,  
Sudan University of Science and Technology,  
PO Box 407, Khartoum, Sudan

Razan Atif  
Research Assistant  
Department of Biomedical Engineering,  
Sudan University of Science and Technology,  
PO Box 407, Khartoum, Sudan

Lina Ahmed  
Research Assistant  
Department of Biomedical Engineering,  
Sudan University of Science and Technology,  
PO Box 407, Khartoum, Sudan

Tahleel Salah Eldeen  
Research Assistant  
Department of Biomedical Engineering,  
Sudan University of Science and Technology,  
PO Box 407, Khartoum, Sudan

Akram Omara  
Assistant Professor  
Department of Biomedical Engineering,  
Sudan University of Science and Technology,  
PO Box 407, Khartoum, Sudan

Megdi Eltayeb  
Assistant Professor  
Department of Biomedical Engineering,  
Sudan University of Science and Technology,  
PO Box 407, Khartoum, Sudan

Abstract—"Cancer" is the most terrifying disease, just hearing the word "C" is a horror for every patient. Notably, the conventional doses form of taking the anticancer drugs are having ruthless side effects and high toxicity on the patient's body. The design of nanoparticles while maintaining the sink condition in the laboratory release trials is one of the biggest challenges and difficulties in design, especially those low-solubility drugs that are used as anticancer agent carriers. If the doctor is aware of the pharmacokinetics of chemotherapy, it will facilitate and help in determining the amount of dosage required in order to obtain a high therapeutic accuracy and reduce toxicity and affect healthy cells. For this purpose, this paper aims to simulate the pharmacokinetic of different types of chemotherapeutic agents to study the kinetics based on solid lipid drug delivery systems nanoparticles (SLN) by examining drug release from different types of devices such as matrix system and reservoir devices controlled by the diffusion mechanism using preposed and estimated values extracted from literature studies which will help to manufacture nanoparticles that meet specific requirements. In order to determine which device type is optimal and best to use as a drug delivery system to cancerous cells and then examin the effect of particle size of this type, ranging from 10 to 100 nm on the anticancer drug release and decide the best size that give the requiered dose of each chemotherapy drugs that were used in this study, we simulated the release behavior of nanoparticles with nano-sphere geometry using the MATLAB software. As majoour finding of this study, the solid lipid based nanoparticles for drug delivery with matrix solution system type shows prolonged drug release which means that this type of systems have high loading capacity of the various chemotheraputic active compounds and can be use sucessfully as anticancer agent drug delivery.

Keywords—Nanoparticles, Cancer treatment, Drug delivery, Solid lipid, Anticancer drug

I. INTRODUCTION

Cancer is the disease that occurs because of the proliferation of cells in an abnormal and uncontrolled in certain places of the body, which produces different types of cancer, according to the place where this imbalance Werner et al.(2016). As an illustration, the
anomalies and proliferation are induced by the genetically changes that control both the regulatory mechanisms and cell signaling Friberg et al.(2005). The fact is that, our bodies are made of different cells and among these cells are the cancerous cells, but not always able to be active. Which means It is not our end are the cancer disease and the hospital. The delivery of the drug to the cancer cells in particular is important to improve the effect of drugs where the release of the drug directly to the cell disease only, without affecting the healthy cells, which contributes significantly to reduce the toxic side effects and increase the efficiency of the drug in the target location. In the light of taken the drug by traditional methods and a schedule for administering the drug which may be changed in the treatment trip based on the needs to repeat the dose or changed if you see more than one doctor Friberg et al.(2005; Jain(2005), which may significantly affect the outcome of the treatment as a result of the increased dose as these types of anti-cancer often have a narrow therapeutic window. Hence, the urgent need to develop drug delivery systems to control drug release based on nanoparticles, additionally using of new techniques that play an important role in effective treatment and also represents a fundamental goal of cancer treatment Bhatia(2016; da Silva Luz et al.(2016). One of the major challenges that scientists have sought over the years to overcome and break their restrictions is the response of the organism to drugs, which is one of the types of general responses to the various stressors and environmental impacts in a constantly changing environment Vizirianakis(2014). Thus, for decades the attention and direction was prominent for pharmaceutical research is to modify drug delivery systems and to seek the development of new dosage forms to overcome drug inefficacy and poisoning that lead to the failure of pharmacological treatment through the development of nanoparticle delivery systems as it contributes to providing the required therapeutic amount of medication to the appropriate location In the body to release it immediately and then maintain its desired concentration Negut et al.(2017; Thassu Deepak et al.(2007). In the first place the main objective of the study is to attracted the attention of interested researchers in this wonderful branch of nanotechnology which known as drug delivery system by examining drug release from different types of nanoparticles based-solid lipid as drug delivery system devices such as matrix system and reservoir devices controlled by the diffusion mechanism and then determin the release profiles for the best device type that is optimal to use as a drug delivery system to cancer cells.

II. THEORY

A. Nanotechnology-based drug delivery systems

Innovation and developments of the new technologies are a result of the urgent need to find new and effective methods of detecting and treating various diseases with less effort, cost and in the first place in short time, where the time factor is a big challenge to various diseases for the reason of, when they are detected in early time and early stage, the treatment will be more easily and efficiently and this is the major lacking in traditional methods Gharpure et al.(2015; Jabir et al.(2012). In the light of these needs in recent years, nanotechnology has gained the attention of researchers, where this technology points out as, the science that allows the study of material treated on the molecular scale and atomic material that ranges from less than 100 nanometers Yousaf et al.(2008). These technologies also allow manipulation, control, study and manufacture of devices with different structures in the range of nanometers Singh et al.(2009). As a key point, nanotechnology can be classified into several categories, which include their different uses due to their nanoscale size, which is not only used in the biomedical field, but also in delivering the drug to the target location Sahoo et al.(2003; Stylios et al.(2005). In fact, nanotechnology researches expanding and focusing especially in the areas of use in the delivery of drugs and genes, where this aspect has become increasingly important in the medical field as the drugs are characterized by the specieification and effectiveness where the spatial and temporal control of the drug. In addition to, prolong the duration of the release of the drug Muhamad12 et al.(2014; Zhang et al.(2013).

Figure 1: Scheme of Nanostructures (A) Nanospheres (matrix system) and (B) Nanocapsules (reservoir system).
In general, nanoparticles, which have recently emerged as an effective carrier technique in the delivery of drugs, can be defined as nanostructures with a scale less than 10 to 1,000 micrometers. The most widely used nanoparticles include polymers, liposomes, quantum dots, proteins, micelles, dendrimers and nanotube in which a drug is either encapsulated or conjugated to internal domains of the carrier Singh et al. (2009). Nanoparticles are a collective term called for both nanospheres (solid matrix) and nanocapsules (reservoir/core-shell), which are the two types commonly used depending on the nanoparticles preparation processes as sown in (Figure 1).

1) **Solid lipid nanoparticles as drug delivery systems**

Fatty nanoparticles are one of the main systems widely used to deliver efficient ingredients because of their ability to protect them from unfavorable environmental conditions which will lead to increases storage levels and releases active components at a constant rate, in addition improving bioavailability within the body Eltayeb et al. (2013). The lipid nanoparticles have two of the most commonly used particles type in used as drug delivery system based-solid lipid nanoparticles (SLN) which have many applications as illustrated in (Figure 2) and nanostructured lipid carrier (NLC) where the SLN consist of solid lipid rather than liquid lipid while the NLC consist of both types of lipid Naseri et al. (2015).

Because SLN is less toxic than polymeric nanoparticles, it is useful as a drug bearer to treat tumors because of their biological and biodegradable nature. In addition, they are able to prolong drug release after taking the dose. This means that the loading and incorporation of anti-cancer drugs into SLN to target tumors will be more effective and less toxic, and will release the appropriate dose Ruckmani et al. (2006; Shenoy et al. (2005).

III. **METHOD**

When talking about controlled drug delivery systems, it is important to describe the mechanisms that controlled and governed the drug release process. It is worth mentioning that a number of physical and chemical phenomena actually affect the release rates such as e.g., drug dissolution, a combination of water diffusion, drug diffusion, polymer/ lipid dissolution, polymer/ lipid swelling, and/or polymer/ lipid degradation Torchilin (2011). In this study, we focused on the diffusion process as it is one of the most extensive mechanisms that control the release of drugs from delivery systems. With attention to the mathematical approaches of diffusion mechanisms for drug release this study was conducted in order to examine the drug release kinetics based on solid lipid drug delivery systems nanoparticles (SLN) by examining drug release from different types of devices such as matrix system and reservoir devices controlled by the diffusion mechanism as shown in (Figure 3).

**Figure 3:** Schematic presentation of classification of the diffusion-controlled drug delivery systems types.

### A. Monolithic Devices (Matrix-Based Solid Lipid Drug Delivery Systems)

These systems are characterized by drug molecules being homogeneous, which means that the drug is dispersed or dissolved though out the matrix that
controlling the release Heller(1987). It is important to know both the ratio or the concentration of the initial drug loaded in the device and the solubility of the drug which facilitate the differentiation between both monolithic solutions and monolithic dispersions and consider by which mechanism the drug is release. The matrix-based system can be classified based on drug solubility and initial drug concentration ratio, as follow:

3.1.1 Monolithic Solutions

In the monolithic solutions matrix, the drug is released by diffusion out of a lipid matrix and the release rate when the initial drug concentration above solubility limit in matrix. Fick's laws of diffusion can be used in order to quantify diffusional behaviour through this devices type J. Siepmann & Siepmann(2012):

$$\frac{M_t}{M_\infty} = 1 - \frac{6}{\pi^2} \sum_{n=1}^{\infty} \exp\left[-\frac{Dn^2\pi^2 t}{R^2}\right]$$

Where R represents the radius of the sphere.

1) Monolithic Dispersions

A single dispersion model where the drug is released by diffusion and controls the release rate by the solubility limitation of the homogeneous drug in the matrix Fan et al.(1989). They also have an initial concentration of drugs that significantly exceeds their solubility. Specifically, in this system part of the drug is dissolved in the matrix while the remainder is dispersed as non-dissolved (crystalline and / or amorphous) J. Siepmann, Siegel, et al.(2012). On the other side, the dispersion of the dissolved drug only can be rapidly replaced by the dissolution of the solid drug. The following equation can be used to decribed the release rate of this system F. Siepmann et al.(2006):

$$\frac{M_t}{M_\infty} = 4\pi R^2 \sqrt{\frac{2C_t - C_e}{C_e}} C_e D t + \frac{4C_e D t}{9R} \left( \frac{C_t}{2C_t - C_e} - 3 \right)$$

where and represent the absolute cumulative amounts of drug released, t is the time, D is the drug diffusion coefficient , R is the spherical device radius, are the initial drug concentration and is the drug solubility within the system.

B. Reservoir Devices (Reservoir-Based Solid Lipid Drug Delivery Systems)

The reservoirs system which called core-shell-structure delivery systems which represent one of the most common drug delivery systems in addition to the matrix systems. First thing to remember, these systems are controlled by the properties of the drug-encapsulated substance, which represent the essence of this drug as its composition, molecular weight and thickness, as well as the chemical and physical properties of the drug, such as the size of the drug molecules, solubility, and molecular weight Langer(1990). These systems are usually taken advantage of for one of the following applications Freiberg et al.(2004):

- In cases of targeted areas and difficult access through the systemic administration.
- Doses that require a long-term cycle and are highly toxic as cancer treatments.
- Medications that require medium or long term administration and are localized to any particular area of the body.
- Drug depot for long-term systemic management.

The reservoir-based system can be classified based on drug solubility and initial drug concentration ratio, as follow:

1) Reservoir System with Non-constant Activity Source

In the case of reservoirs with non-constant source of activity, the drug encapsulated within core structure which is completely separated from the controlled release rate substance which forms the particle shell which surrounding the drug store and forms a barrier membrane Yang et al.(2012). In other words and as an illustration, the drug solubility is greater than the initial drug concentration . In other words, the molecules of the drug that are administered through the membrane are not replaced, so the concentration of the drug on the surface of the internal membrane gradually decreases over time Langer(1990). The drug diffusion through reservoir device with a “nonconstant activity source” to controlling the release rate via membrane can illustrated by the equation J. Siepmann & Siepmann(2012):

$$\frac{M_t}{M_\infty} = 1 - \exp\left(-\frac{3R_0 D K t}{R_1^2 R_0 - R_1^2}\right)$$

where and represent the cumulative drug released at time t and infinity, respectively, D is the diffusion coefficient of the drug within the membrane, K is the partition coefficient of the drug between the membrane and
the reservoir and are the inner and outer radius of the device.

2) Reservoir System with Constant Activity Source

In the case of reservoirs with constant source of activity, the drug concentration become greater than drug solubility. Consequently, this system have limited solubility where only part of the drug is dissolved and the drug molecules that were released is fastly replaced by dissolution of the excess and remaining of non-dissolved drug in the matrix Yang et al.(2012). Thus, Fick’s law of diffusion equation can be used to describe the release rate from spherical geometry:

\[
M_t = \frac{4\pi DN_c s_c R_o R_i^3}{R_o - R_i} t
\]

Where is the solubility of the drug in the core.

IV. RESULT AND DISSECTION

In this section the MATLAB simulation results of the diffusional mathematical expressions of the drug release profiles and kinetics from SLN system types were represented and described. Markedly, solid lipid nanoparticles drug delivery systems with diffusion controlled devices release mechanism can follow many types of devices (i.e. Matrix Devices which can be monolithic solutions or monolithic dispersion or Reservoir controlled devices with constant or with non-constant activity source etc). To point out, matrix release rate can control by choice of matrix type: (glassy matrix: \(D = 10-10.12\) cm²/s or rubbery matrix: \(D = 10-6.10-7\) cm²/s). In the monolithic matrix the drug is released by diffusion out of a lipid matrix and the release rate depends on initial drug concentration. By the same token, if the drug initial concentration \(C_i\) above solubility limit in matrix the drug dissolution in polymer matrix limits release rate (Monolithic Solution) in inverse side if the initial drug concentration is below solubility limit in matrix the diffusion through matrix limits the release rate (Monolithic Dispersions).

Similarly, matrix with monolithic solutions are homogeneous Nano-spheres solutions consisting of a complete spherical matrix characterized by the homogeneity of the drug, which can be immediately dissolved and distributed to the entire spherical matrix. In the case of this matrix system, we set the limit of drug initial concentration \(C_i = 0.08\) g/cm², drug solubility \(C_s = 0.06\) g/cm² and varying the radius as 0.3, 0.75 and 1.20 nm in order to studying the release rate type compared with the reservoir type. From (Figure 4, Figure 5, Figure 6 and Figure 7) can clearly see that the form/type of release followed the first order or exponential release which means that the release kinetics are not zero order. By the same token, as shown in (Figure 4, Figure 5 and Figure 6) the release rate from matrix with monolithic dispersion is very low compare to the monolithic solution systems and the initial drug concentration \(C_i = 0.06\) g/cm² compared with the drug solubility \(C_s = 0.08\) g/cm² is low. The release rate has affected by radius of device, as the value of radius decrease, the release rate increase and follow a constant behavior.

From (Figure 4, Figure 5 and Figure 6), the ratio of the drug released over time from reservoir system is low compare to the matrix one. Since this system is greatly affected by spherical particle device inner and outer radius, i.e., when the core-shell structure ratio was 2:1 and 3:1 for the delivery device formulations F1, F2 and F3, respectively it is noticeable to observe that when the core: shell ratio is higher, the larger release rate, which mean that the larger the size of the core-structure as in F2 was twice greater than the shell, the greater the amount of drug loaded in the device. In this case of the reservoir device is characterized by a constant activity source that the drug release kinetics follow the zero-order release and the drug is released at a constant rate i.e. the concentration gradient of the drug within the membrane is constant. Moreover, because the drug is dispersed in the excretions, which is the basis of the particle, when the shell has a small thickness or the core: shell ratio is small, it facilitates water penetration of the system, which causes the molecules to dissolve quickly, so the dissolution of the drug compared to the diffusion of the drug is fast, resulting in the release of small amounts in this case for the Doxorubicin chemotherapeutic agent over 21 days. While the chemotherapy agents drug need sustained release over days (continuous infusions) by diffusion while this type of release characterized by the low level of the drug by diffusion over time, which are considered as big challenge of this type of treatment Wyss Institute for Biologically Inspired Engineering at Harvard(2014). We found that, the solid lipid based nanoparticles for drug delivery with matrix solution type compared with other types as represented in (Figure 4, Figure 5, Figure 6 and Figure 7) shows prolonged drug release which means that this type of systems has high loading capacity of the various chemotherapy active compounds which means this systems type offers a novel way to locally deliver the drugs to the targeted place.
Table 1: Chemotherapeutic Agent examples and its application

<table>
<thead>
<tr>
<th>Chemotherapeutic Agent</th>
<th>Anti-cancer Diffusion coefficient (cm²/s)</th>
<th>Application</th>
<th>Partition coefficients</th>
<th>Dose/Days</th>
<th>Reference</th>
</tr>
</thead>
<tbody>
<tr>
<td>Doxorubicin</td>
<td>6.7 x 9</td>
<td>Breast cancer</td>
<td>3.91</td>
<td>60 to 75 mg/m² repeated every 21 days</td>
<td>Alves et al. (2017); Prescribers Digital Reference (2018); Weinberg et al. (2007)</td>
</tr>
<tr>
<td>Taxanes</td>
<td>2.56 x 12.6</td>
<td>Ovarian cancer</td>
<td>40.03</td>
<td>60-100 mg/m² repeated every 21 days</td>
<td>Markl et al. (2017)</td>
</tr>
<tr>
<td>Cisplatin</td>
<td>2.1 x 29.5</td>
<td>Bladder cancer</td>
<td>0.0865</td>
<td>50 to 70 mg/m² repeated every 28 days</td>
<td>Panczyk et al. (2013); Suresh Gyan Vihar University (2013)</td>
</tr>
<tr>
<td>Doxil</td>
<td>3.72 x 32.5</td>
<td>Cell Lymphoma cancer</td>
<td>1.34</td>
<td>20 to 40 mg/m² repeated every 28 days</td>
<td>Toley et al. (2013)</td>
</tr>
</tbody>
</table>

Table 2: Base parameters values for used in doxorubicin drug and delivery devices for simulation (Dose 60 mg/m² in 21 days).

<table>
<thead>
<tr>
<th>Formulation</th>
<th>Diffusion Coefficient (x cm²/s)</th>
<th>Radius nm</th>
<th>Maximum Release Rate</th>
</tr>
</thead>
<tbody>
<tr>
<td>F1</td>
<td>6.7</td>
<td>9</td>
<td>89.65</td>
</tr>
<tr>
<td>F2</td>
<td>6.7</td>
<td>12.6</td>
<td>80</td>
</tr>
<tr>
<td>F3</td>
<td>6.7</td>
<td>26</td>
<td>62</td>
</tr>
<tr>
<td>F4</td>
<td>6.7</td>
<td>29.5</td>
<td>60</td>
</tr>
<tr>
<td>F5</td>
<td>6.7</td>
<td>32.5</td>
<td>58.5</td>
</tr>
</tbody>
</table>

Table 3: Base parameters values for used in Taxotere drug and delivery devices for simulation (Dose 60-100 mg/m² in 21 days).

<table>
<thead>
<tr>
<th>Formulation</th>
<th>Diffusion Coefficient (x cm²/s)</th>
<th>Radius nm</th>
<th>Maximum Release Rate</th>
</tr>
</thead>
<tbody>
<tr>
<td>F1</td>
<td>2.56</td>
<td>2</td>
<td>65</td>
</tr>
<tr>
<td>F2</td>
<td>2.56</td>
<td>12.6</td>
<td>57</td>
</tr>
<tr>
<td>F3</td>
<td>2.56</td>
<td>26</td>
<td>42.50</td>
</tr>
</tbody>
</table>

Table 4: Base parameters values for used in cisplatin drug and delivery devices for simulation (Dose 50-70 mg/m² in 28 days).

<table>
<thead>
<tr>
<th>Formulation</th>
<th>Diffusion Coefficient (x cm²/s)</th>
<th>Radius nm</th>
<th>Maximum Release Rate</th>
</tr>
</thead>
<tbody>
<tr>
<td>F1</td>
<td>2.1</td>
<td>2</td>
<td>67</td>
</tr>
<tr>
<td>F2</td>
<td>2.1</td>
<td>12.6</td>
<td>59.50</td>
</tr>
<tr>
<td>F3</td>
<td>2.1</td>
<td>26</td>
<td>44</td>
</tr>
</tbody>
</table>

Table 5: Base parameters values for used in Doxil drug and delivery devices for simulation (Dose 20-40 mg/m² in 28 days).

<table>
<thead>
<tr>
<th>Formulation</th>
<th>Diffusion Coefficient (x cm²/s)</th>
<th>Radius nm</th>
<th>Maximum Release Rate</th>
</tr>
</thead>
<tbody>
<tr>
<td>F1</td>
<td>3.72</td>
<td>52.5</td>
<td>41.88</td>
</tr>
<tr>
<td>F2</td>
<td>3.72</td>
<td>60.5</td>
<td>39.5</td>
</tr>
<tr>
<td>F3</td>
<td>3.72</td>
<td>70</td>
<td>37</td>
</tr>
<tr>
<td>F4</td>
<td>3.72</td>
<td>83.2</td>
<td>34.1</td>
</tr>
<tr>
<td>F5</td>
<td>3.72</td>
<td>90.5</td>
<td>33</td>
</tr>
</tbody>
</table>
In this study we compared different matrix solution spherical lipid nanoparticles delivery system loaded with various chemotherapy agent drugs such as (Doxorubicin, Taxanes, Cisplatin and Doxil) as listed in (Table 1) and the diffusion was simulated using (Eq.1) with difference particle radius which was ranging from 10 to 100 nm as listed in (Table 1 and 3).

In table 4 and 5 we found that as illustrate in (Figure 8, Figure 9, Figure 10 and Figure 11) the release rate amount over time is affected inversely by particle size, whereas the spherical geometric radius increases the release rate decreases and in the case of doxorubicin drug the amount of required dose is in the range (60 to 75 mg/m²) at 21 days, as listed in (Table 2) the values of radius 26 and 29.5 nm are suitable to give the required amount when the diffusion coefficient D is cm²/s as shown in (Figure 8), for Taxanes drug the amount of required dose is in the range (60 to 100 mg/m²) at 21 days, as listed in table 3, the values of radius 9 nm is suitable to give the required amount when the diffusion coefficient D is cm²/s as shown in (Figure 9), also for Cisplatin drug the amount of required dose is in the range (50 to 70 mg/m²) at 28 days, as listed in table 4, the values of radius 9 and 12.6 nm are suitable to give the required amount when the diffusion coefficient D is cm²/s as shown in (Figure 10), finally in the case of Doxil the amount of required dose is in the range (20 to 40 mg/m²) at 28 days, as listed in (Table 5) all the values of radius are suitable to give the required amount when the diffusion coefficient D is cm²/s as shown in (Figure 11).

In other words, the active ingredient release from the nanoparticle its clear in that it is very sensitive to this particle size Eltayeb et al.(2015). Markedly, many factors are able to govern the drug release from the nanoparticles inside the body after administration and those factors include drug solubility, drug initial concentration, matrix and drug diffusion coefficients, particle size and partition coefficient Yahia et al.(2018) and from the data set that were reported in the Eltayeb et al.(2016) experiment where this study worked with both homogenous matrix and core-shell nanoparticles with tenable size we deduced that also the polymer to drug ratio is effected on the particle delivery system tape as well as the drug release as 1:1 or 4:4 w%, the system tends to be a matrix particle, on the same way when the polymer to drug ratio is 1:3, 1:4 or 2:4 w%, the system tends to be core-shell particle.
V. CONCLUSION

The solid lipid nanoparticles occupy the top list of nanoparticles in the treatment of cancer, although they have a few side effects, they are expected to have prominence in the future in the field of cancer treatment for human beings. The release of drug molecules from both matrix systems or reservoir devices can be predicted using mathematical equations to describe drug release kinetics with varying properties of drug-encapsulated substances, altering the chemical and physical properties of the drug and analyze their effects on the launch process where the release rate can be simulated easily by using computer software. In addition, these programs can be used to better understand the proliferation mechanism that controls drug release from advanced delivery systems. Significantly, it is important to know both the ratio or the concentration of the initial drug loaded in the device and the solubility of the drug which facilitate the differentiation between the matrix and reservoir delivery devices system types. As well as know the radius and size of nanoparticles which is a major factor that affecting the drug release and from the computational model we found that the drug release could be controlled by varying the radius values for the formulations.

VI. REFERENCE


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